

# The Chemistry of Salinity in Soils

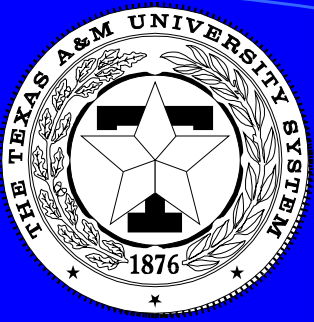
Naomi Waissman Assadian

Texas A&M Research and Extension

Center at El Paso

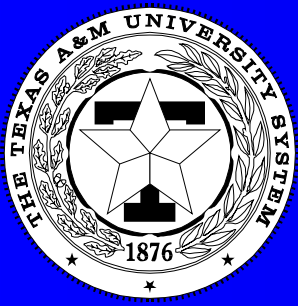
September 4, 2002





# Introduction

- **Salinity concentrations in soils have profound effects on soil properties**
  - **Poor drainage, nutrient deficiencies, and crop toxicity**
- **Factors in soil salinity include:**
  - **Arid climates**
  - **Erosion of primary minerals**
  - **Saline irrigation and drainage waters**
  - **High water table**
  - **High potential Evapotranspiration**
  - **Additions of inorganic and organic fertilizers**
  - **Sludges and sewage effluents**



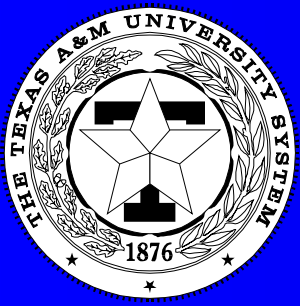
“The term **Salinity** refers to the presence of the major dissolved inorganic solutes, essentially

$\text{Na}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{K}^+$ ,

$\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{HCO}_3^-$ ,  $\text{CO}_3^{2-}$  “

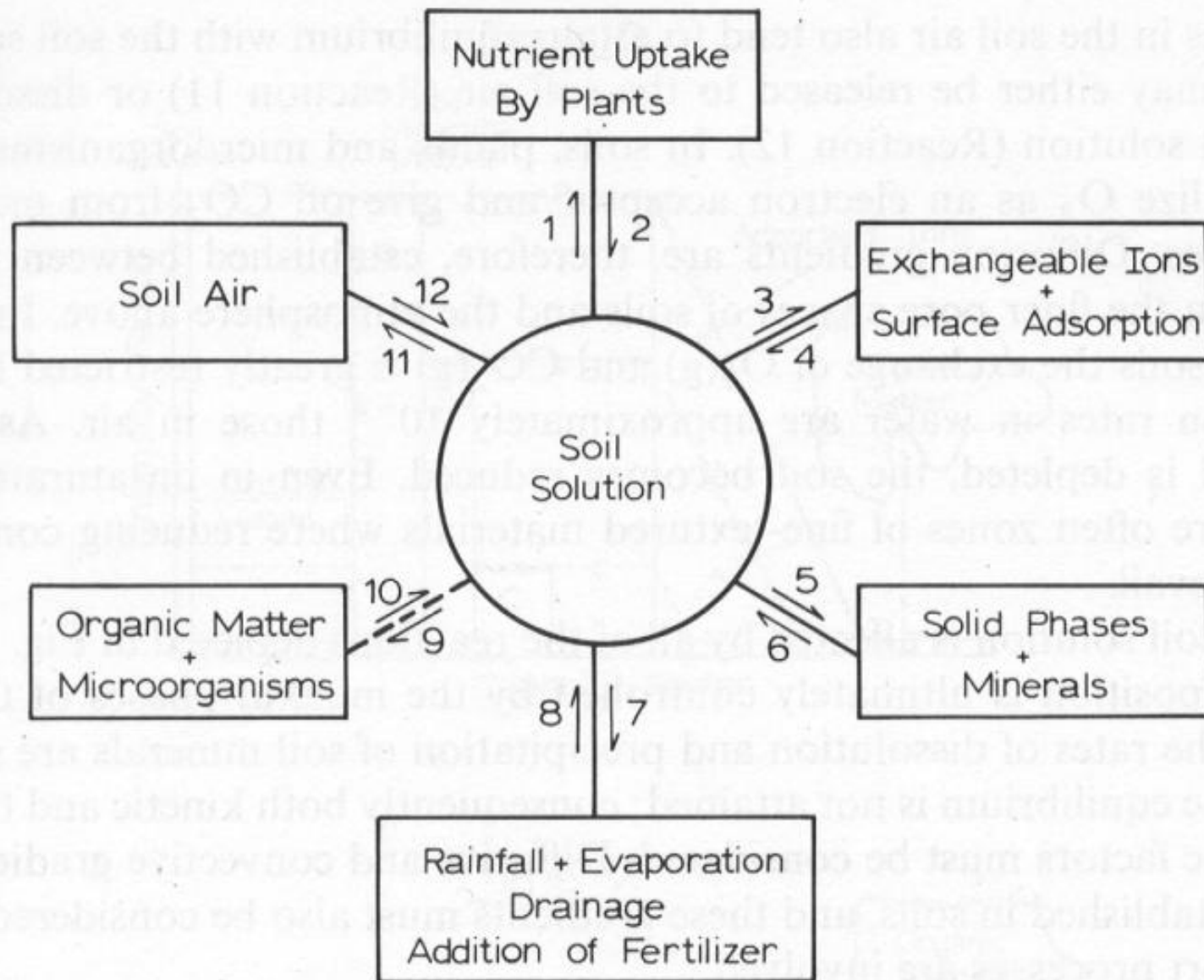
SSSA BOOK SERIES: 5, METHODS OF SOIL ANALYSIS PT 3 – CHEMICAL METHODS.

(It could also include  $\text{NH}_4^+$ ,  $\text{NO}_3^-$ ,  $\text{PO}_4^{3-}$ , and charged organic particles in aqueous (soil) samples when determined by electrical conductance)

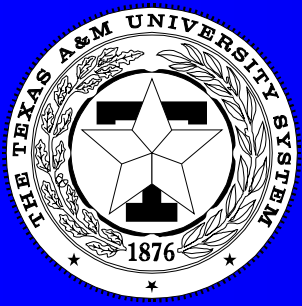


# Dissolved Solids (Salts)

- **Intensity Factor** – salt concentration in soil solution
- **Capacity Factor** – the ability of solid phases to replenish elements as it is depleted from solution



**Fig. 1.1** The dynamic equilibria that occur in soils.

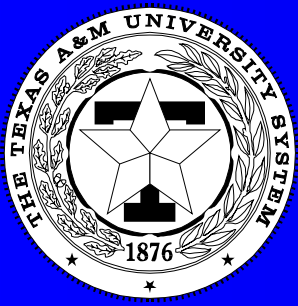


# Salt Measurement – Intensity Factor

- **Electrical Conductance** – EC – Electrical current carried by a salt solution. EC increases as salt concentrations increase.

## Soil Characteristics

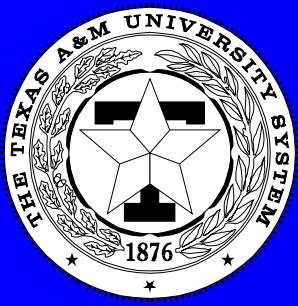
- pH is generally below 8.5
- EC is more than 4 dS/m



# Calculations

- Total Dissolved Solids= $(\sim 640)(\text{EC})$
- Ionic Strength= $(0.013)(\text{EC})=\sum c_i Z^2$
- Single-Ion Activity
  - $\alpha_j = \gamma_j m_j$   $m$ =molar concentration
- Extended Debye Huckle Equation (estimate activity coefficients)
  - $\text{Log} \gamma = -AZ^2[(I)^{1/2}/(1+B\text{\AA}(I)^{1/2})]$ 
    - $A=0.512$  for water @  $25^\circ\text{C}$ ,
    - $B\sim 0.33$  in water @  $25^\circ\text{C}$
    - $\text{\AA}$ =adjustable parameter corresponding to ion size and charge (6)

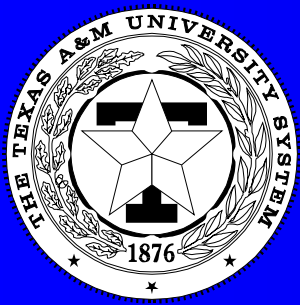




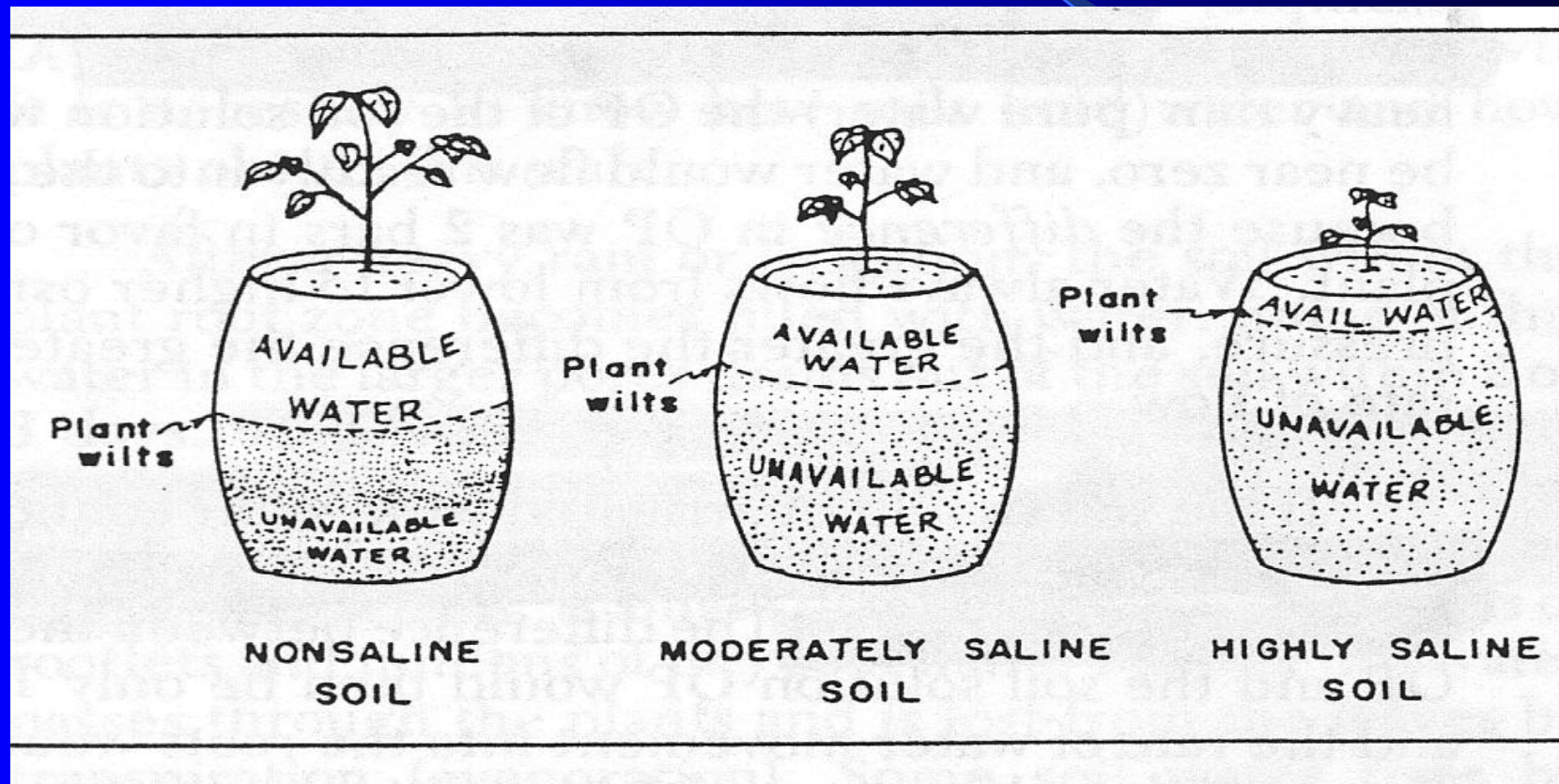
# Factors Affecting Salt Dissolution

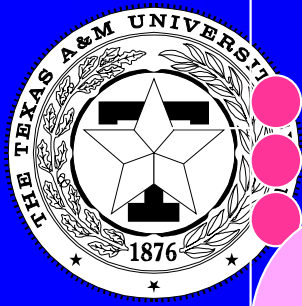
- **Salt Effect or Ionic Strength Effect** – Addition of an inert salt can increase the solubility of an ionic compound
  - Added NaCl can increase  $\text{CaSO}_4$  dissolution
  - Increased ionic strength affects pH
- **Ion pairing effect**
  - Weak bonds between hydrated ions
- **Common Ion Effect** – a salt will be less soluble if one of its constituent ions is already present in the solution



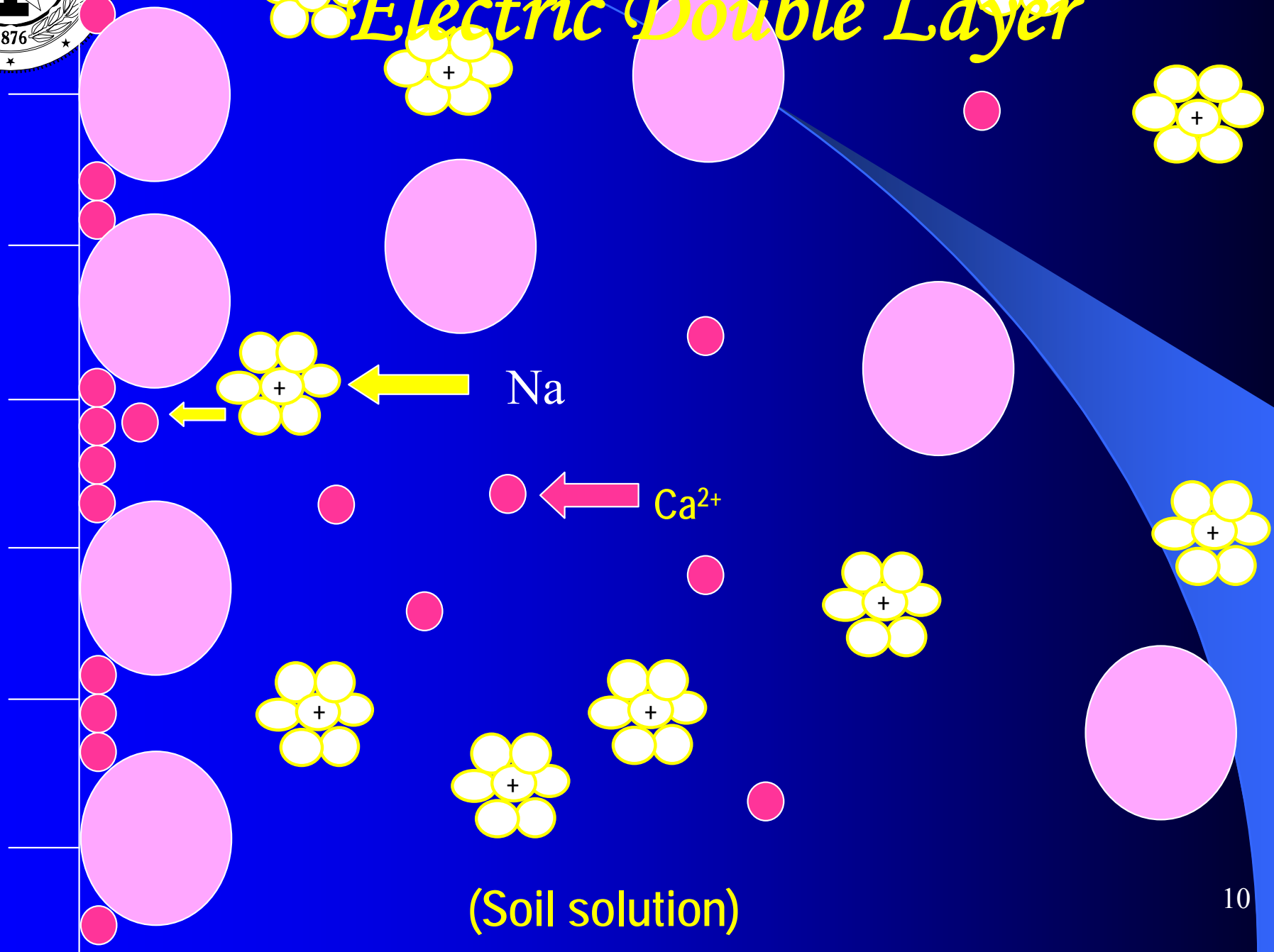


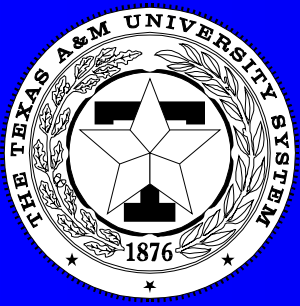
# Osmotic Potential (kPa) = 40EC





# The Physiochemical Effect *The Electric Double Layer*





Sodium Adsorption Ratio (SAR): Defined as being “related to the adsorption of sodium by the soil”

$$SAR = Na^+ / (Ca^{++} + Mg^{++}) / 2$$

Na<sup>+</sup>, Ca<sup>++</sup>, and Mg<sup>++</sup> in milliequivalent/L (Allison, 1954)

➤ **Sodicity** occurs when the ratio of sodium is much higher to that of other soluble salts.

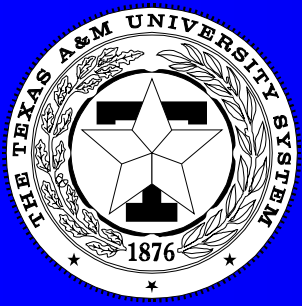
### **Soil Characteristics**

➤ pH ranges between 8.5 to 10,  
ESP > 15

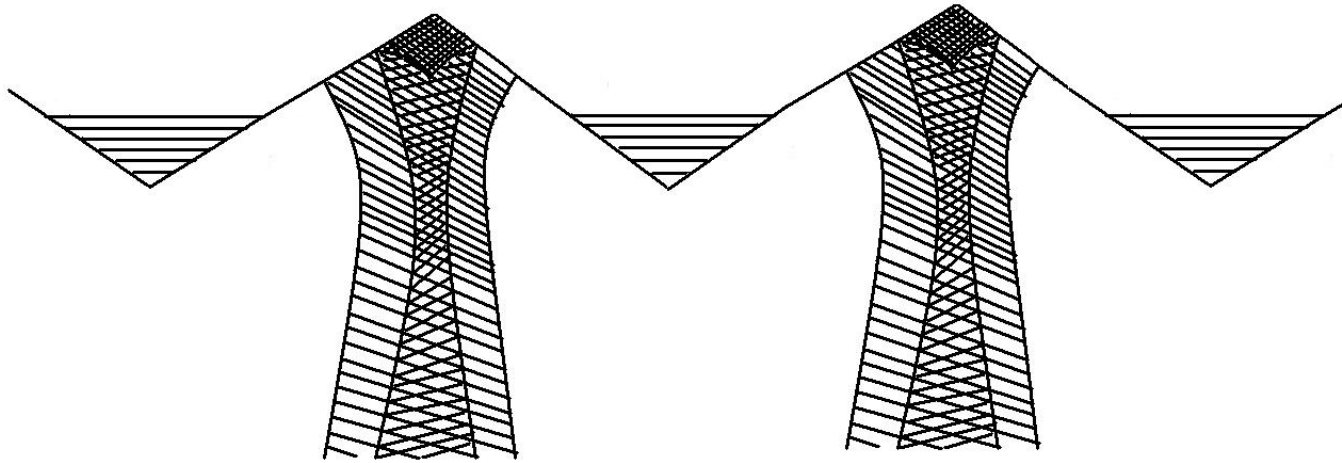
➤ Plastic and sticky when wet, forming  
hard clods and crusts

➤ Impermeable to all water.

• **Specific Ion Effect** – Na ion can be inhibitory  
to certain plant processes



# Surface Salt Pattern



Low salt  
accumulation



Moderate salt  
accumulation



High salt  
accumulation

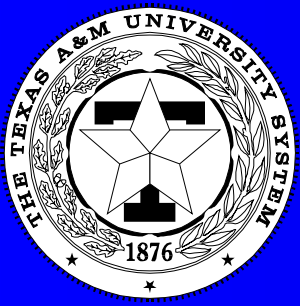


Very high salt  
accumulation

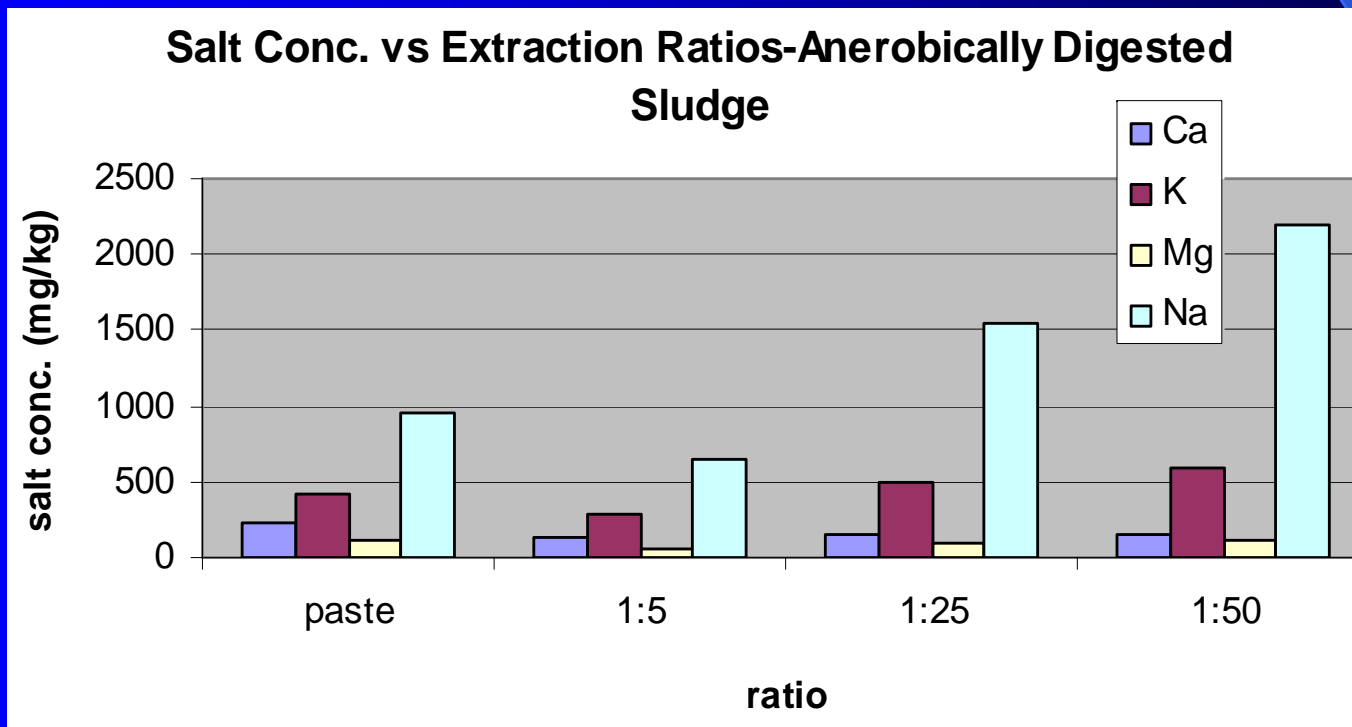
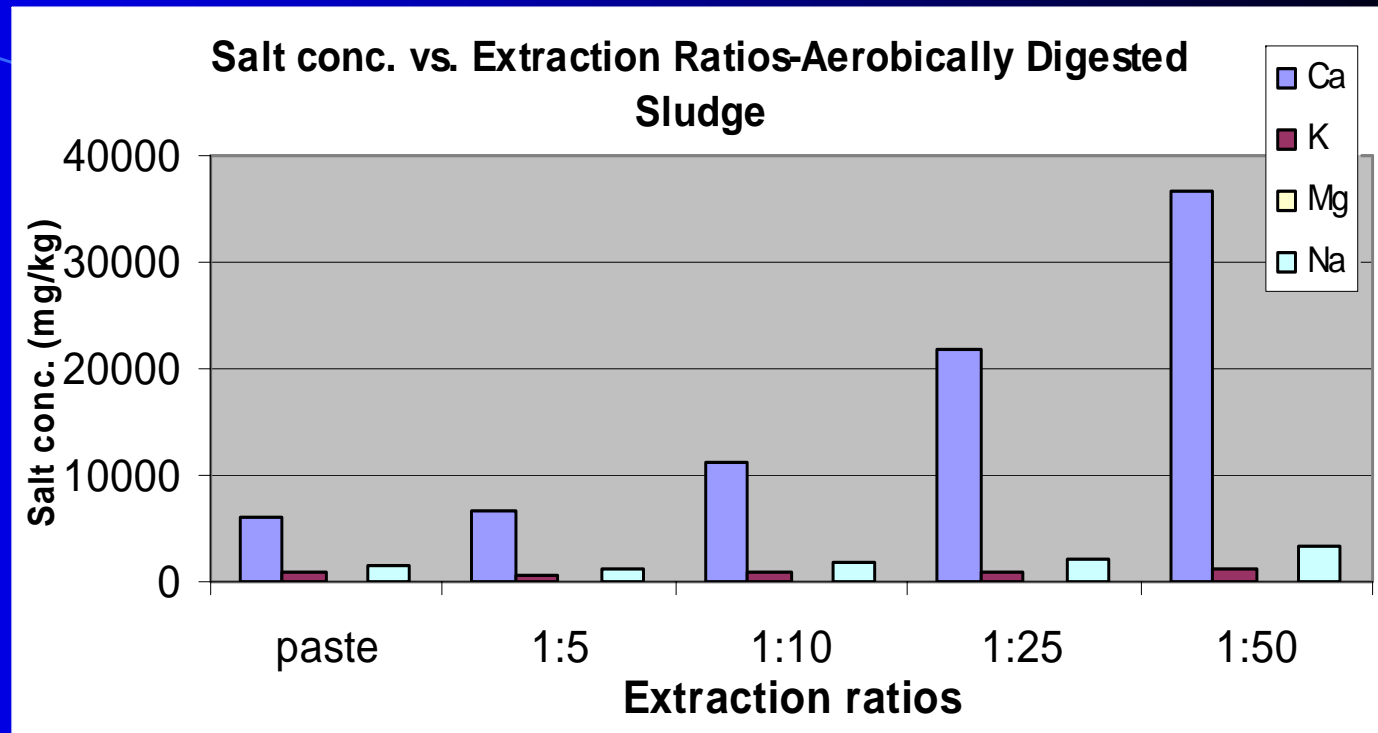




# Current Research

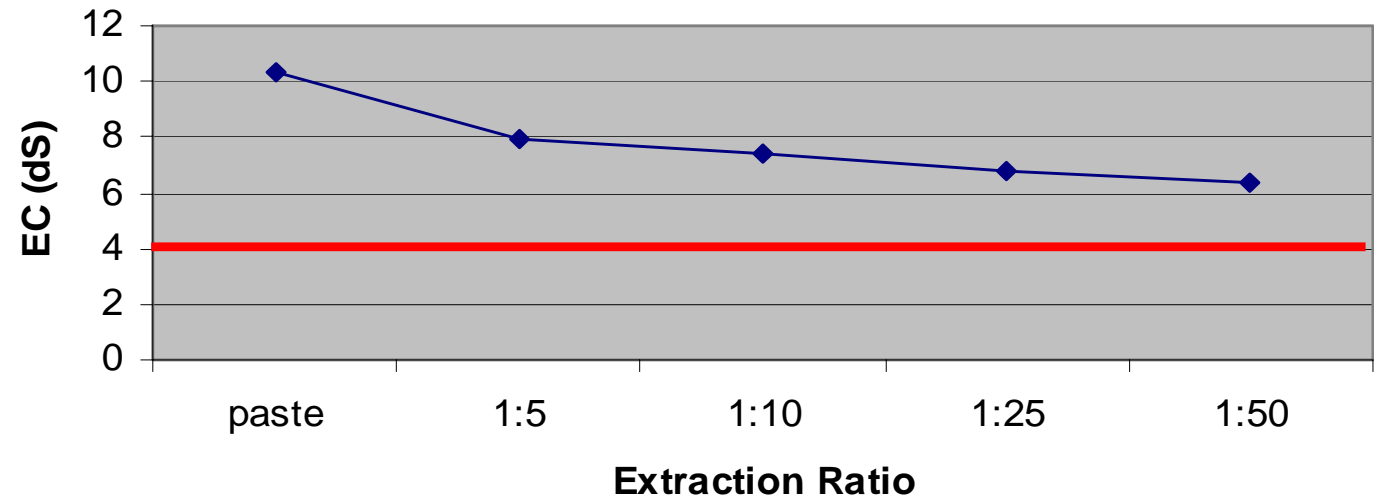


## Study #1

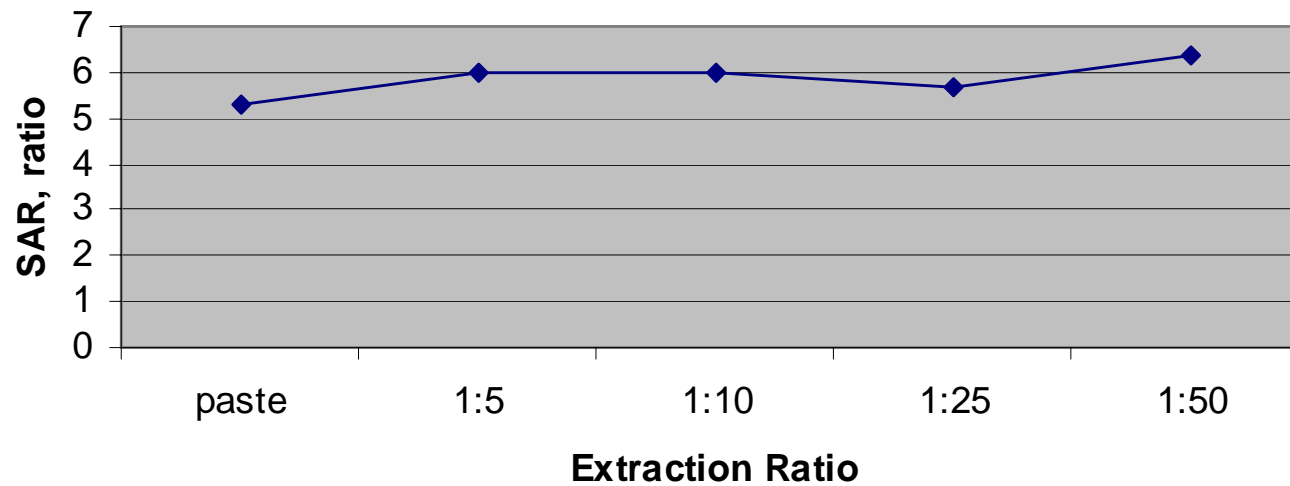




## Extraction Ratio's impacts on EC-Aerobically Digested Sludge



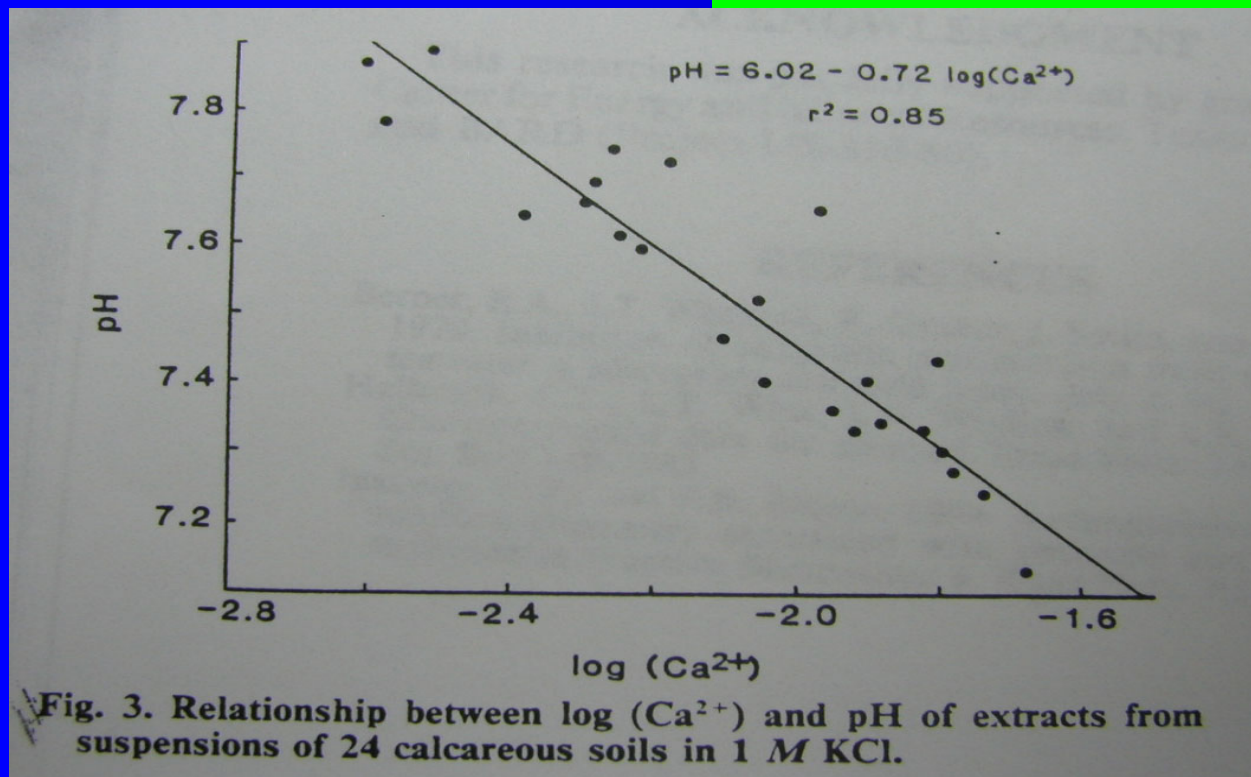
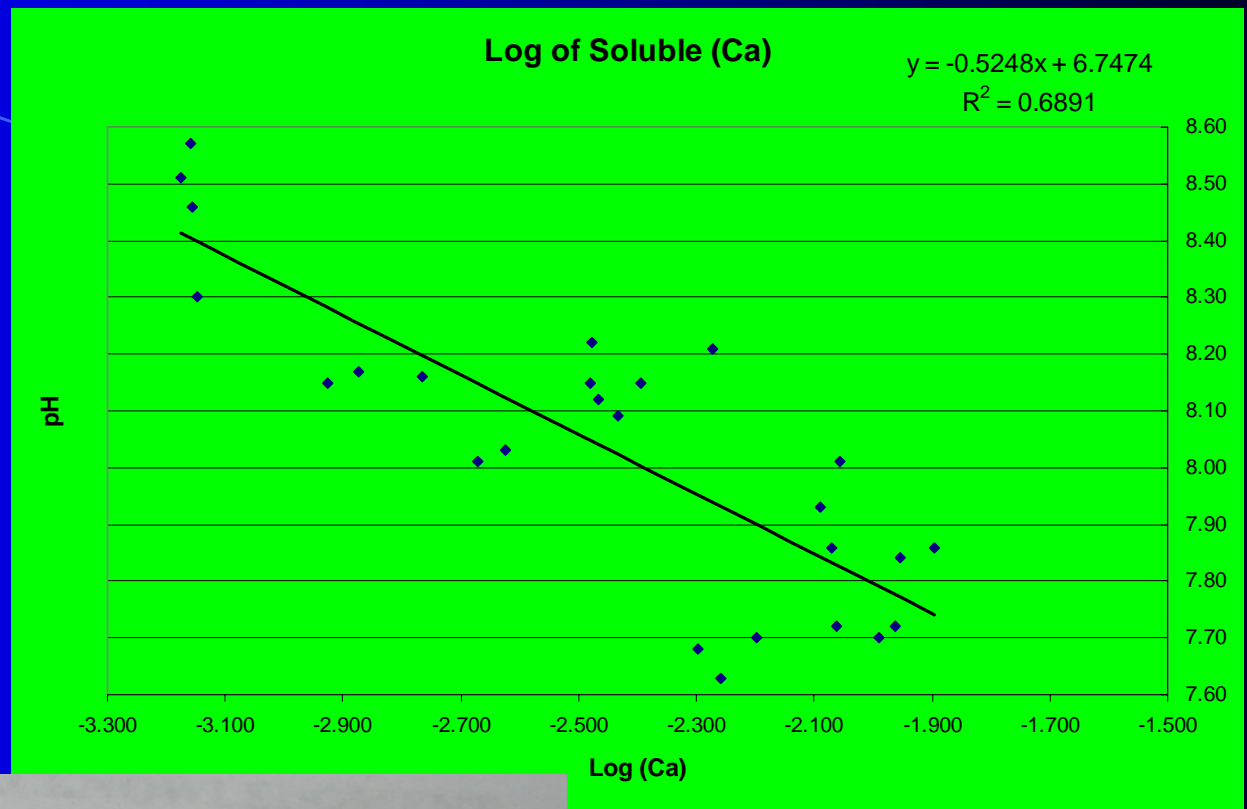
## Sodium Adsorption Ratio-Aerobically Digested Biosolids

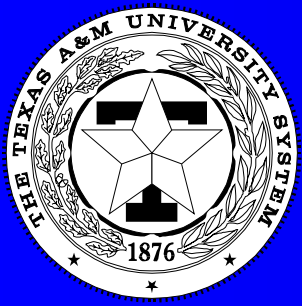






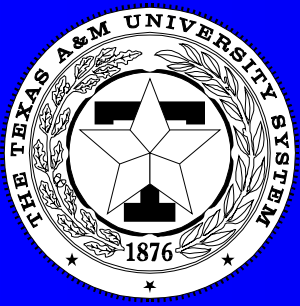
## Study #2





# Study #3 - Objectives

- ❖ Determine if minimum tillage is possible on irrigated soils with a high potential for salt hazard.
- ❖ Determine if biosolids remediate accumulation of salts at the soil surface.
- ❖ Evaluate the impact of ammonium on soil salinity.



# Treatment Design

## Tillage

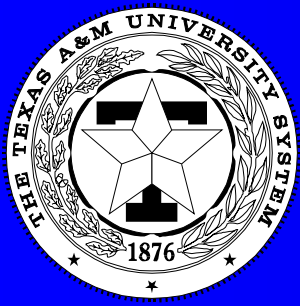
- ❖ Maximum
- ❖ Minimum

## Soil Texture

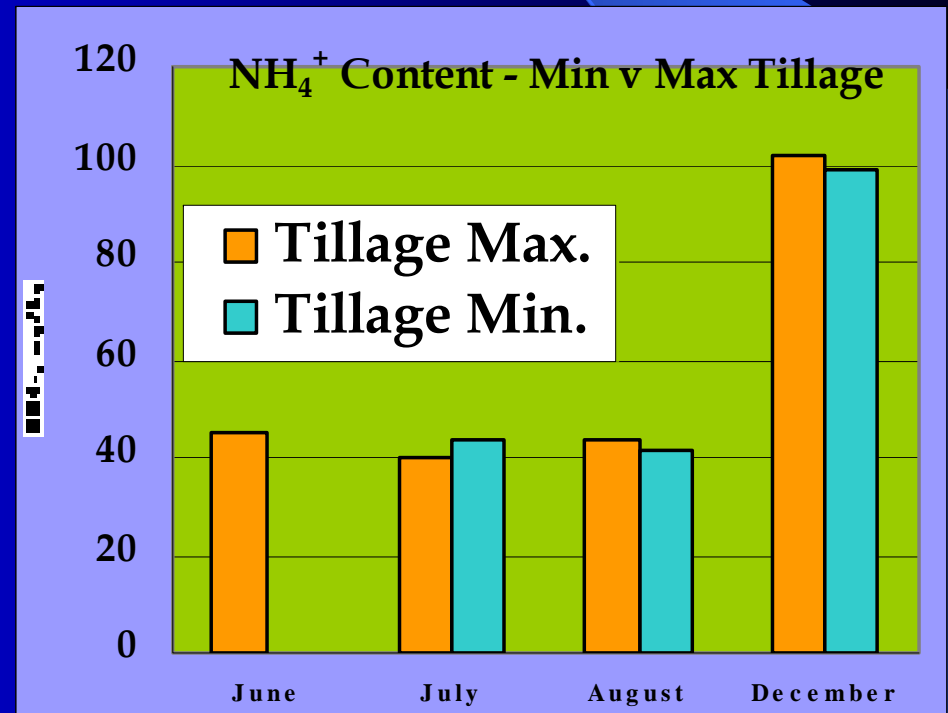
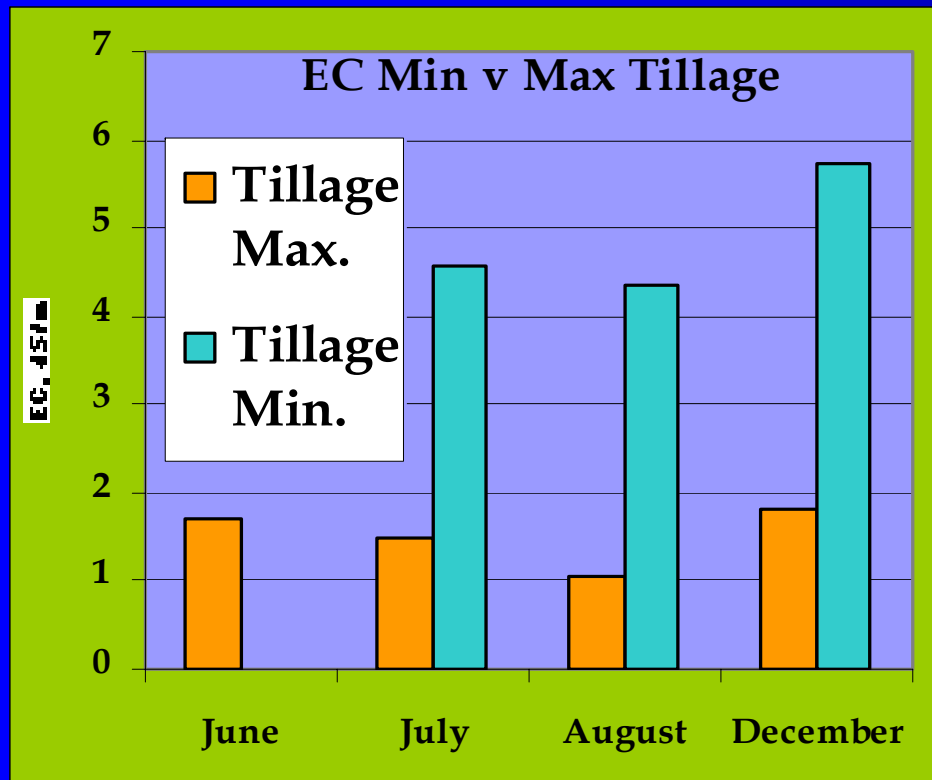
- ❖ Ge – Glendale silty clay loam.
- ❖ Gs – Glendale silty clay.
- ❖ Hs – Harkey silty clay loam.
- ❖ Tg – Tigua silty clay.

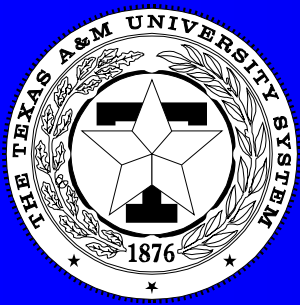
## Subplots – Bed Row Position

- ❖ Furrow
- ❖ Side
- ❖ Top
- ❖ Center

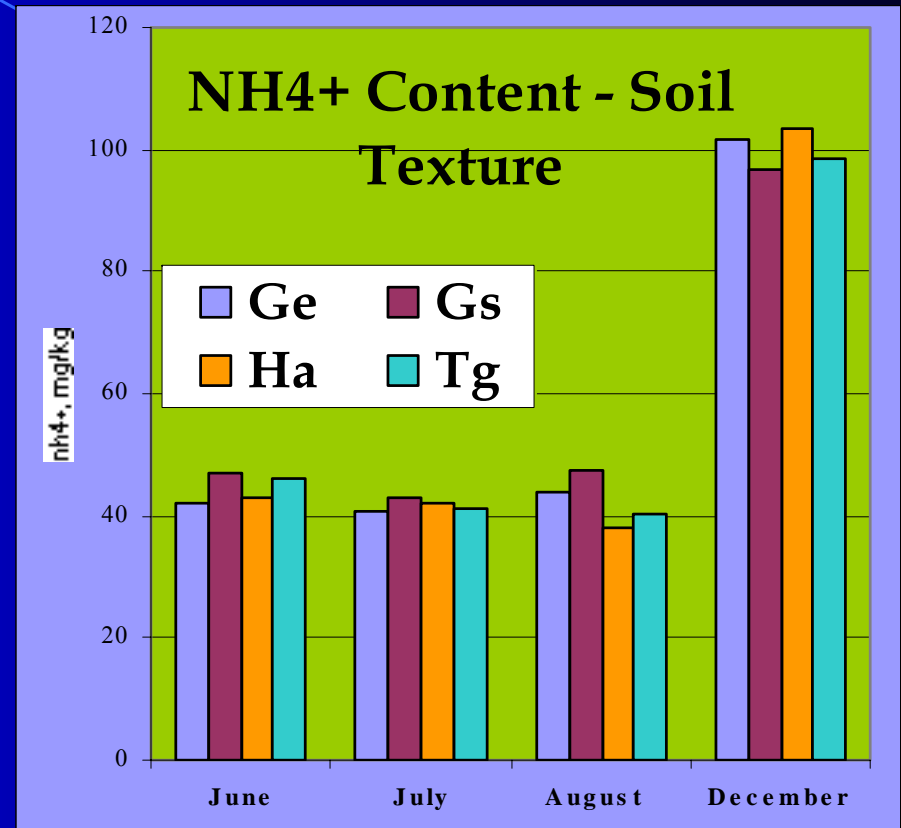
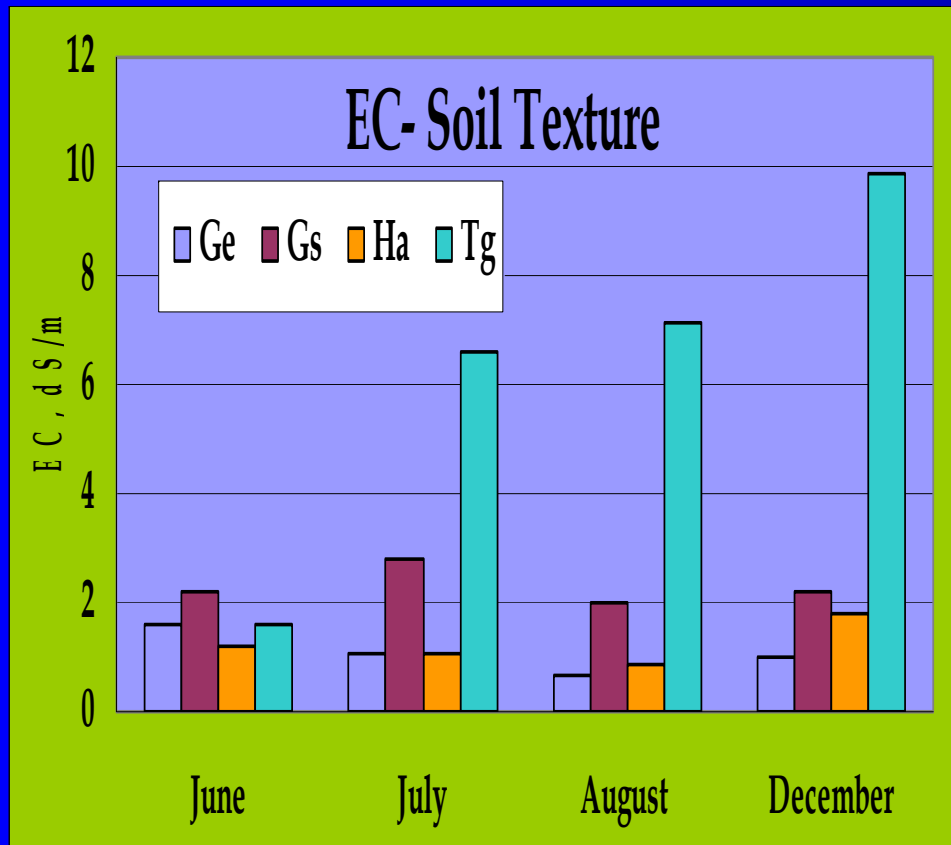


# Minimum versus Maximum Tillage



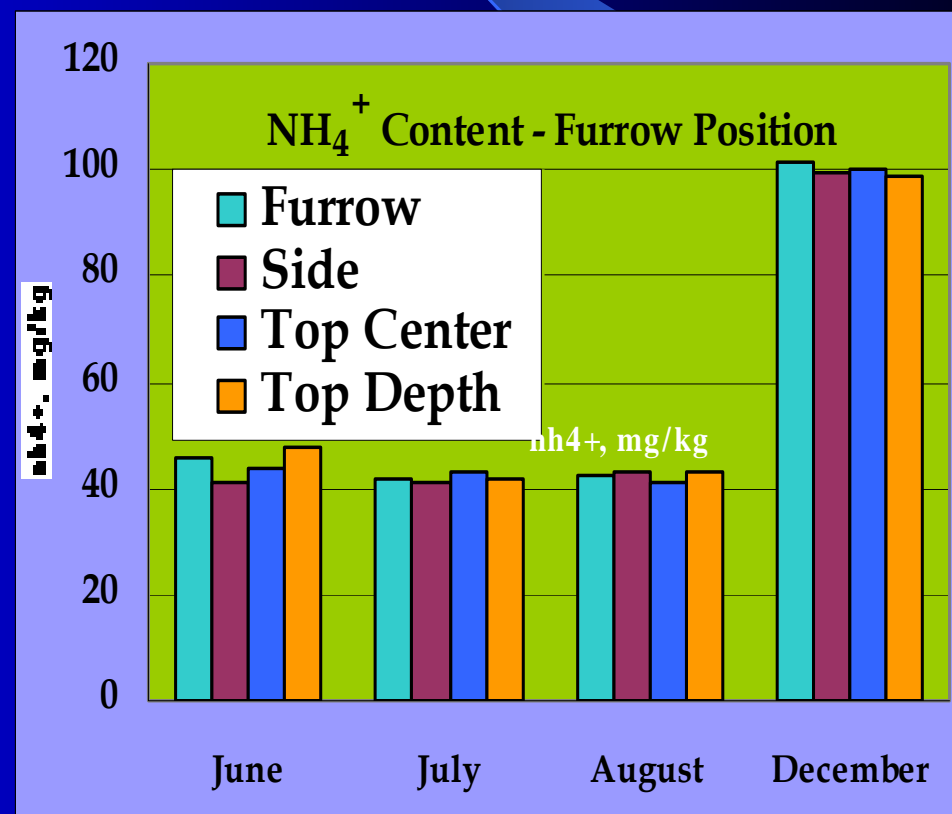
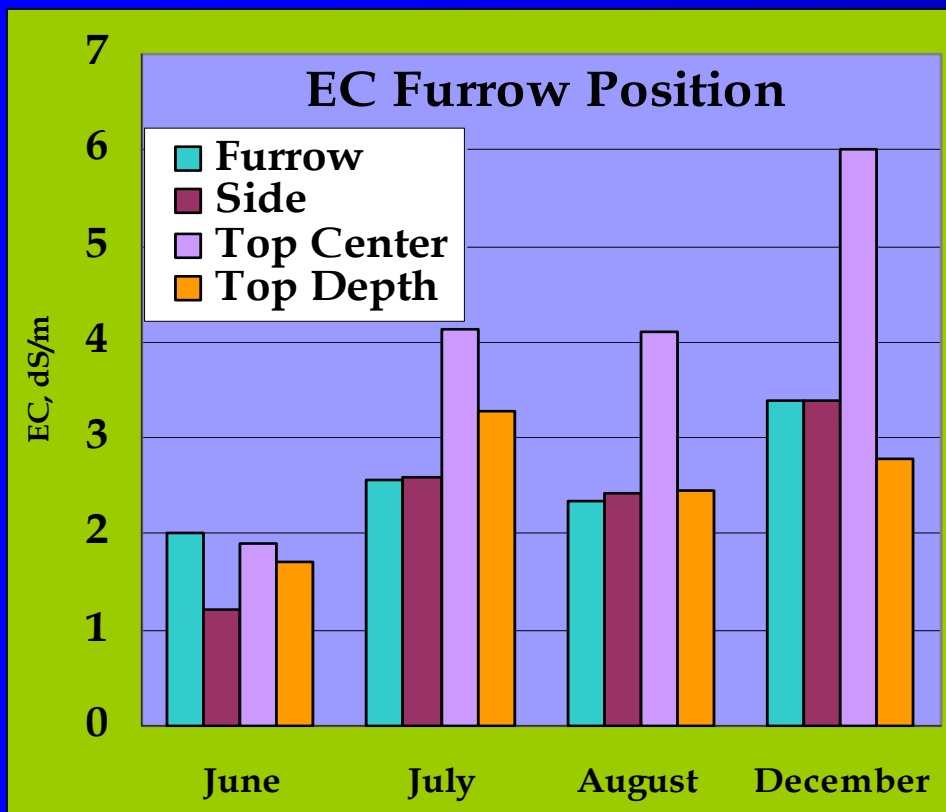


# Soil Texture





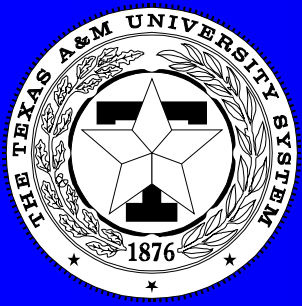
# Furrow Position





# The Future in the Upper Rio Grande Basin





# Potential Salt Management Strategies

- Develop a consistent and detailed salt monitoring program
  - Miyamoto and Iglesias
- Crop Selection – Halophytes
  - Miyamoto and USSL
- Pretreatment of alternative water sources for irrigation
  - Relatively new, site-specific
- Irrigation delivery systems
  - Enciso
- Integrated Soil Management to maintain soil structure and permeability (organic enrichment)